This article was downloaded by: On: *25 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Liquid Crystals

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713926090

Preliminary communication First example of a functionalized triphenylene discotic trimer: molecular engineering of advanced materials Sandeep Kumar; M. Manickam

Online publication date: 06 August 2010

To cite this Article Kumar, Sandeep and Manickam, M.(1999) 'Preliminary communication First example of a functionalized triphenylene discotic trimer: molecular engineering of advanced materials', Liquid Crystals, 26: 6, 939 – 941

To link to this Article: DOI: 10.1080/026782999204642 URL: http://dx.doi.org/10.1080/026782999204642

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Preliminary communication

First example of a functionalized triphenylene discotic trimer: molecular engineering of advanced materials

SANDEEP KUMAR* and M. MANICKAM

Centre for Liquid Crystal Research, P.O. Box 1329, Jalahalli, Bangalore—560 013, India

(Received 28 December 1998; accepted 13 January 1999)

A new type of functionalized discotic oligomer system has been realized whose molecular architecture consists of a trisubstituted benzene as central core and three nitro-functionalized triphenylenes as regional cores. These regional cores are linked to the central core via ether linkages through an alkyl chain spacer. Polarizing microscopy and calorimetric measurements reveal the existence of a monotropic columnar mesophase in this novel functionalized oligomer. The product does not crystallize at room temperature over a long period (c. five months) or on cooling to -50° C. The corresponding unfunctionalized trimer is not liquid crystalline. The dipolar interaction of nitro groups is probably responsible for the induction of mesogenicity. The introduction of functionality into liquid crystal oligomers permits the variation of their properties on a wide scale and opens the route to new synthetic supramolecular systems for various device applications.

The work of Chandrasekhar on the hexa-esters of benzene resulted in the discovery of discotic liquid crystals (LCs) [1]. Since then, a number of hexasubstituted benzene derivatives have been reported to show columnar mesophases. These include various hexa-alkanoyloxybenzene derivatives, hexakis(alkoxyphenoxymethyl) derivatives, hexakis(alkylsulfone) derivatives, various multiynes, and a combination of ester and amide groups [2]. Some partially acylated and tetrasubstituted benzene derivatives are also known to show columnar mesophases [2]. A few 1,3,5-trisubstituted benzene derivatives were also reported to form columnar mesophases but their structures have not been confirmed unambiguously [2]. It was anticipated that three elongated chains are not sufficient to fill the space around the core and therefore, could not allow the existence of columnar mesophases [3]. In order to fill the space around the trisubstituted cores, branched substituents have been designed, to create 'super-disk' cores [4] with either six or nine elongated side chains [3, 5]. Only a few compounds having only three side chains, e.g. decacyclene derivatives [6], and triphenylene derivatives [7] are known to show columnar mesophases.

Triphenylene derivatives play an important role in the field of discotic LCs. These thermally and chemically

*Author for correspondence; e-mail: uclc@giasbg01.vsnl.net.in

stable materials can easily be prepared and purified. The very high charge carrier mobility in various mesophases of these materials makes them suitable for use in one-dimensional conductors [8, 9], photoconductors [9, 10] and light-emitting diodes [11]. One-dimensional energy transfer has also been studied extensively in these materials [12].

In general, there are four main physical parameters to control in all opto-electronic LC device configurations: (1) supramolecular order of the mesophase, (2) stability of the mesophase, (3) processability, and (4) photophysical and electronic properties of the LC material. A combination of these properties in a compound requires design and synthesis of novel functionalized discotic dimers, oligomers, networks or polymers. Functionalization of the core is necessary for tailoring the mesophase properties, photophysical properties and processability; several research groups are therefore currently working on the synthesis of functionalized triphenylene derivatives [13]. Previously, we have shown that various hydroxyalkoxytriphenylenes are accessible by simple strategies, often on a large scale [13, 14]. In an effort to enhance fluorescence as well as the mesogenicity of triphenylene discotic LCs, we have synthesized a number of new monofunctionalized triphenylene and mixed tail triphenylene discotics bearing conjugative electron withdrawing substituents attached directly to the triphenylene core

[15]. In order to induce a molecular dipole, colour, etc., Boden and Praefcke and their co-workers have recently nitrated and halogenated the triphenylene core [16, 17]. We have recently observed that not only the one ring as reported [16, 17] but all the three rings of hexaalkoxytriphenylene can be nitrated easily [18]. Hitherto, only these two types of functionalized discotic materials were known. We realized that hybridization of these two varieties of functionalized discotics would lead to extremely important novel double functionalized derivatives, where one type of functional group can be used to tune the electronic nature of the molecule and, at the same time, the other may be utilized to convert the molecules to a processable dimer, oligomer or polymer. Very recently we have reported the first examples of functionalized discotic dimers [19]. In this preliminary communication we wish to report the first example of a functionalized discotic oligomer.

The synthesis of the (nitro) functionalized triphenylene oligomer is shown in the scheme. The functionalized

C₄H₀C

C₄H

ОН

Y OC₄H₀ OC₄H₀

CH₂Cl₂ / Et₃N

OC₄H₉

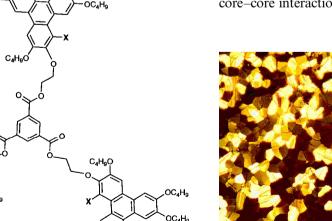
ore [16, 17].alcohol with one molecule of 1,3,5-benzenetricarboxylic
acid chloride easily furnished the desired functionalized
trimer 2a. It was purified by column chromatography
over silica gel and well characterized by spectral analysis.
The corresponding unfunctionalized trimer 2b was also
prepared similarly, for comparison of the mesophase
behaviours.would lead
unctionalized
roup can be
ecule and, atThe thermal behaviour of both trimers was investi-
gated by polarizing optical microscopy and differential
scanning calorimetry (DSC) on 1–2 mg samples using a

scanning calorimetry (DSC) on 1-2 mg samples using a Perkin-Elmer DSC7 with heating and cooling rates of 10°C min⁻¹. The functionalized trimer shows two peaks on first DSC heating run: a weak transition at 161.1°C $(\Delta H, 2.0 \text{ Jg}^{-1})$ and another broad peak at 181.3°C with a large ΔH (36.3 J g⁻¹), corresponding to a transition from the crystalline solid state to the isotropic liquid state. Upon cooling from isotropic liquid the columnar phase is formed at 158.7°C (ΔH , 10.9 J g⁻¹). The optical texture of this mesophase (figure 1) closely resembles the known textures for Col_h phases and it is miscible with the well characterized Colh phase of mononitrohexa-alkoxytriphenylene [16, 18]. On second heating it shows a sharp peak at 162.0°C (ΔH , 10.6 J g⁻¹). The mesophase reappears on cooling at 158.8°C and does not show any sign of crystallization on keeping at room temperature for more than five months. Dimers and oligomers often show glass transitions at low temperature but we did not observe any clear glass transition or crystallization on cooling to -50° C. On the other hand, the corresponding unfunctionalized oligomer 2b is a non-liquid crystalline material. It melts at 185°C and crystallizes at 166°C on cooling.

triphenylene alcohol 1a was prepared as reported

earlier [18]. Condensation of three molecules of this

In these oligomers, strong interactions of the triphenylene units is rather more important than the core–core interaction of the central benzene core, which



 OC_4H_9 **b**: X = H OC_4H_9 Scheme: Synthesis of trimers 2a and 2b.

a: X = NO₂

CAHOC

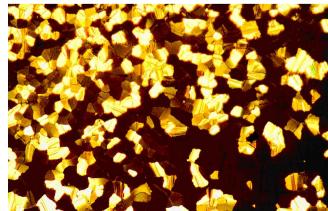


Figure 1. Optical texture of the oligomer 2a at 30°C.

C+H_oC

C₄H₉C

C₄H₅

is negligible as reflected by the long spacing (4.7 Å) between the benzene rings in trisubstituted benzene derivatives [20]. As the unfunctionalized trimer 2b is non-mesogenic, the dipolar interaction of the nitro group is possibly of importance in the supramolecular structure formation of the functionalized trimer 2a. However, several other derivatives with different spacers and alkyl chains must be prepared to confirm this conclusion. Work in this direction is in progress in this laboratory.

Selected data for compound **2a**: Mass: m/z (FAB) 2237.4; ¹H NMR (CDCl₃): δ 9.07 (s, 3H), 7.58 (s, 3H), 7.45 (s, 3H), 7.41 (s, 3H), 7.34 (s, 3H), 7.12 (s, 3H), 4.7 (br s, 6H), 4.5 (br s, 6H), 4.29 (t, 6H), 4.19 (t, 6H), 4.06 (t, 12H), 3.71 (t, 6H), 1.9–1.3 (m, 60H), 1.05 (m, 36H) and 0.77 (t, 9H).

Selected data for compound **2b**: Mass: m/z (FAB) 2102.5; ¹H NMR (CDCl₃): δ 8.71 (s, 3H), 7.76 (m, 18H), 4.7 (m, 6H), 4.5 (m, 6H), 4.2 (m, 30H), 1.9–1.3 (m, 60H) and 1.0 (m, 45H).

We are grateful to Prof. S. Chandrasekhar for helpful discussions. Technical assistance of Mr. Sanjay K. Varshney is gratefully acknowledged. We are also thankful to Mr. Holger Schonherr and Prof. G. J. Vancso for providing mass spectra.

References

- [1] CHANDRASEKHAR, S., SADASHIVA, B. K., and SURESH, K. A., 1977, Pramana, 9, 471.
- [2] CAMMIDGE, A. N., and BUSHBY, R. J., 1998, Hand Book of Liquid Crystals, Vol. 2B, edited by D. Demus, J. Goodby, G. W. Gray, H.-W. Spiess and V. Vill (Wiley-VCH), Chap. VII.
- [3] MALTHETE, J., and COLLET, A., 1987, J. Am. chem. Soc., 109, 7644.
- [4] KOHNE, B., and PRAEFCKE, K., 1985, Chem. Ztg., 109, 121.
- [5] LATTERMANN, G., 1987, *Liq. Cryst.*, 2, 723; LATTERMANN, G., and HOCKER, H., 1986, *Mol. Cryst. liq. Cryst.*, 133, 245.
- [6] KEINAN, E., KUMAR, S., MOSHENBERG, R., GIRLANDO, R., and WACHTEL, E. J., 1993, *Liq. Cryst.*, 11, 157.
- [7] HENDERSON, P., KUMAR, S., REGO, J. A., RINGSDORF, H., and SCHUHMACHER, P., 1995, J. chem. Soc., chem. Commun., 1059.

- [8] BODEN, N., and MOVAGHAR, B., 1998, Hand Book of Liquid Crystals, Vol. 2B, edited by D. Demus, J. Goodby, G. W. Gray, H.-W. Spiess and V. Vill (Wiley-VCH), Chap. IX.
- [9] CHANDRASEKHAR, S., 1998, Hand Book of Liquid Crystals, Vol. 2B, edited by D. Demus, J. Goodby, G. W. Gray, H.-W. Spiess and V. Vill (Wiley-VCH), Chap. VIII.
- [10] ADAM, D., CLOSS, F., FREY, T., FUNHOFF, D., HAARER, D., RINGSDORF, H., SCHUHMACHER, P., and SIEMENSMEYER, K., 1993, Phys. Rev. Lett., 70, 457; ADAM, D., SCHUHMACHER, P., SIMMERER, J., HÄUSSLING, L., SIEMENSMEYER, K., ETZBACH, K. H., RINGSDORF, H., and HAARER, D., 1994, Nature, 371, 141; SIMMERER, J., GLUSEN, B., PAULUS, W., KETTNER, A., SCHUHMACHER, P., ADAM, D., ETZBACH, K. H., SIEMENSMEYER, K., WENDORFF, J. H., RINGSDORF, H., and HAARER, D., 1996, Adv. Mater., 8, 815.
- [11] STAPFF, I. H., STUMPFLEN, V., WENDORFF, J. H., SPOHN, D. B., and MOBIUS, D., 1997, *Liq. Cryst.*, 23, 613.
- [12] MARKOVITSI, D., GERMAIN, A., MILLIE, P., LECUYER, P., GALLOS, L. K., ARGYRAKIS, P., BENGS, H., and RINGSDORF, H., 1995, J. phys. Chem., 99, 1005; MARGUET, S., MARKOVITSI, D., MILLIE, P., SIGAL, H., and KUMAR, S., 1998, J. phys. Chem. B, 102, 4697 and references therein.
- [13] KUMAR, S., and MANICKAM, M., 1998, *Synthesis*, 1119 and references therein.
- [14] KUMAR, S., SCHUHMACHER, P., HENDERSON, P., REGO, J., and RINGSDORF, H., 1996, *Mol. Cryst. liq. Cryst.*, 288, 211 and references therein.
- [15] REGO, J. A., KUMAR, S., and RINGSDORF, H., 1996, *Chem. Mater.*, 8, 1402; REGO, J. A., KUMAR, S., DMOCHOWSKI, I. J., and RINGSDORF, H., 1996, *J. chem. Soc., chem. Commun.*, 1031.
- [16] BODEN, N., BUSHBY, R. J., and CAMMIDGE, A. N., 1995, Mol. Cryst. liq. Cryst., 260, 307; BODEN, N., BUSHBY, R. J., and CAMMIDGE, A. N., 1995, Liq. Cryst., 18, 673; BODEN, N., BUSHBY, R. J., CAMMIDGE, A. N., and HEADDOCK, G., 1995, J. mater. Chem., 5, 2275.
- [17] PRAEFCKE, K., ECKERT, A., and BLUNK, D., 1997, *Liq. Cryst.*, **22**, 113; BODEN, N., BUSHBY, R. J., CAMMIDGE, A. N., DUCKWORTH, S., and HEADDOCK, G., 1997, *J. mater. Chem.*, **7**, 601.
- [18] KUMAR, S., and MANICKAM, M., 1998, Mol. Cryst. liq. Cryst., 309, 291; KUMAR, S., MANICKAM, M., BALAGURUSAMY, V. S. K., and SCHONHERR, H., Liq. Cryst. (submitted).
- [19] KUMAR, S., MANICKAM, M., and SCHONHERR, H., *Liq. Cryst.* (submitted).
- [20] MATSUNGA, Y., MIYAJIMA, N., NAKAYASU, Y., SAKAI, S., and YONENAGA, M., 1988, Bull. chem. Soc. Jpn., 61, 207.